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balanced reaction how many moles of  $\text{NO}$  are formed from 8.44 moles of  $\text{NO}_2$  if there is plenty of water present  $3\text{NO}_2 + \text{H}_2\text{O} \rightarrow 2\text{HNO}_3 + \text{NO}$  consider the following reaction how many moles of oxygen are required to produce 2.33 moles of water

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